



Modeling of Variation on Oil Quenching with Iterative Treatment Using Cellular Automaton

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Agenda

- ✓ Background
- ✓ Low dimension cellular automaton method
- ✓ Simulation of heat treatment and cooling variations
- ✓ Deformation analysis consider with heat treatment and cooling variations
- ✓ Conclusion





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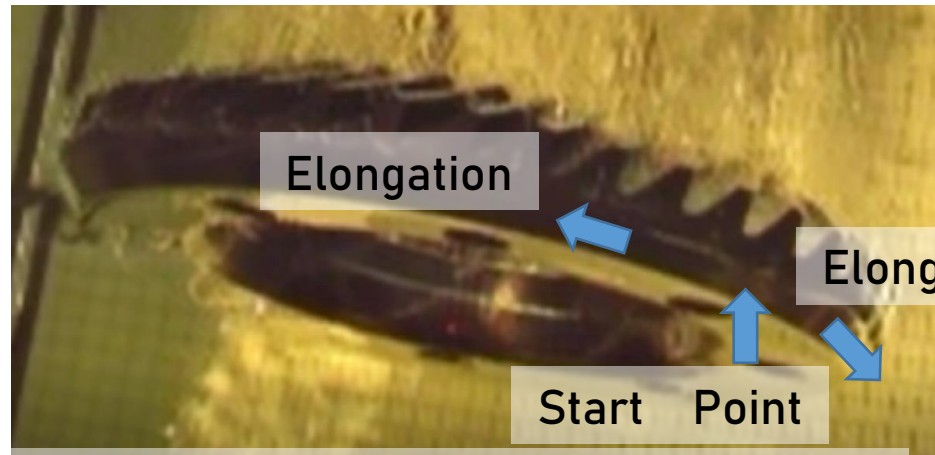
Relationship between Cooling state and Deformation

In heat treatment simulation of oil quenching process, it is necessary to solve the vapor blanket/boiling/convection each stage and add boundary conditions to the part model surface.

Require expensive calculation cost
Difficulties in complex shape

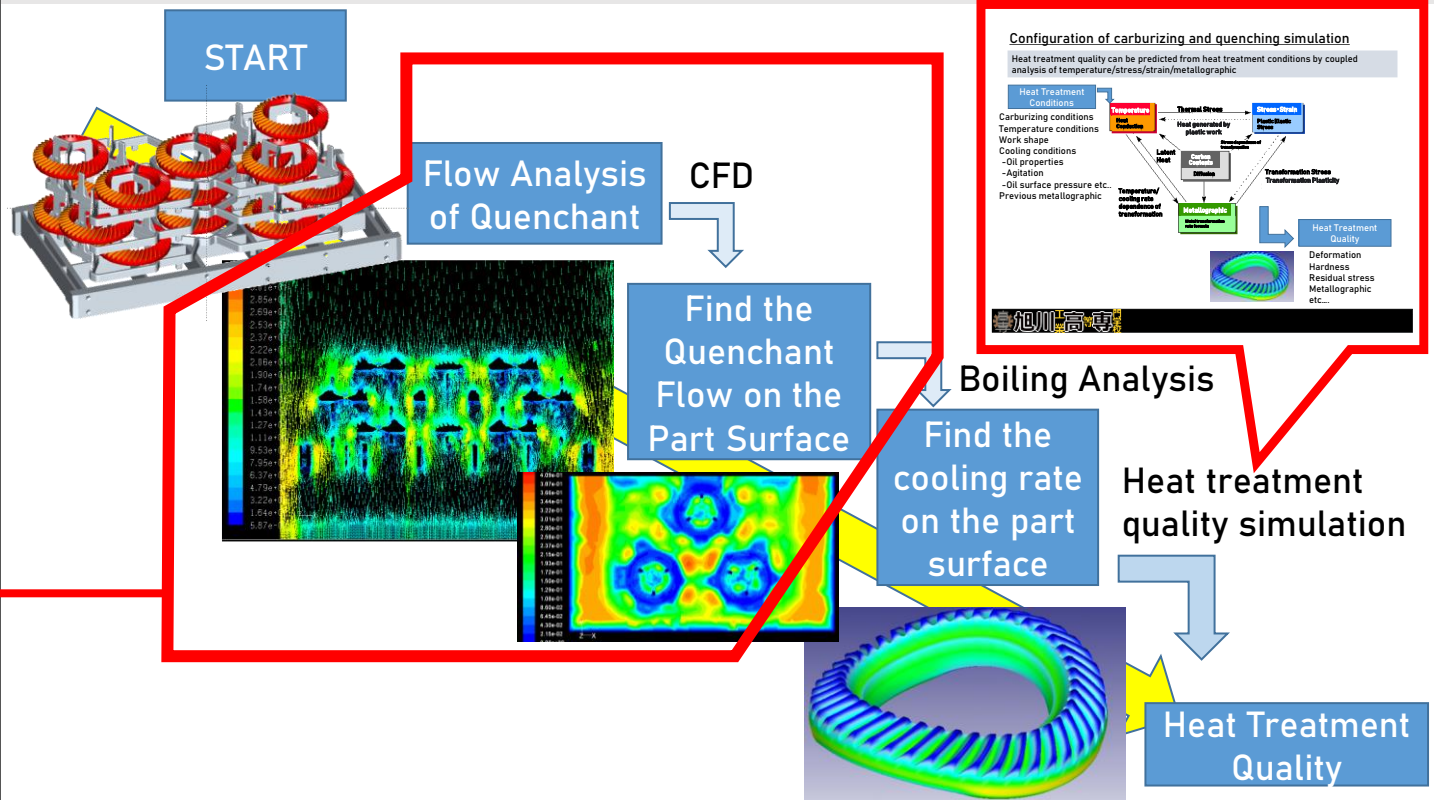
“Low dimensional cellular automaton”

Visualization of vapor film collapse mode during unsteady boiling on oil quenching by using cellular automaton simulation, Tsuyoshi Sugimoto, 27th International Federation for Heat Treatment and Surface Engineering, Salzburg, Austria 2022



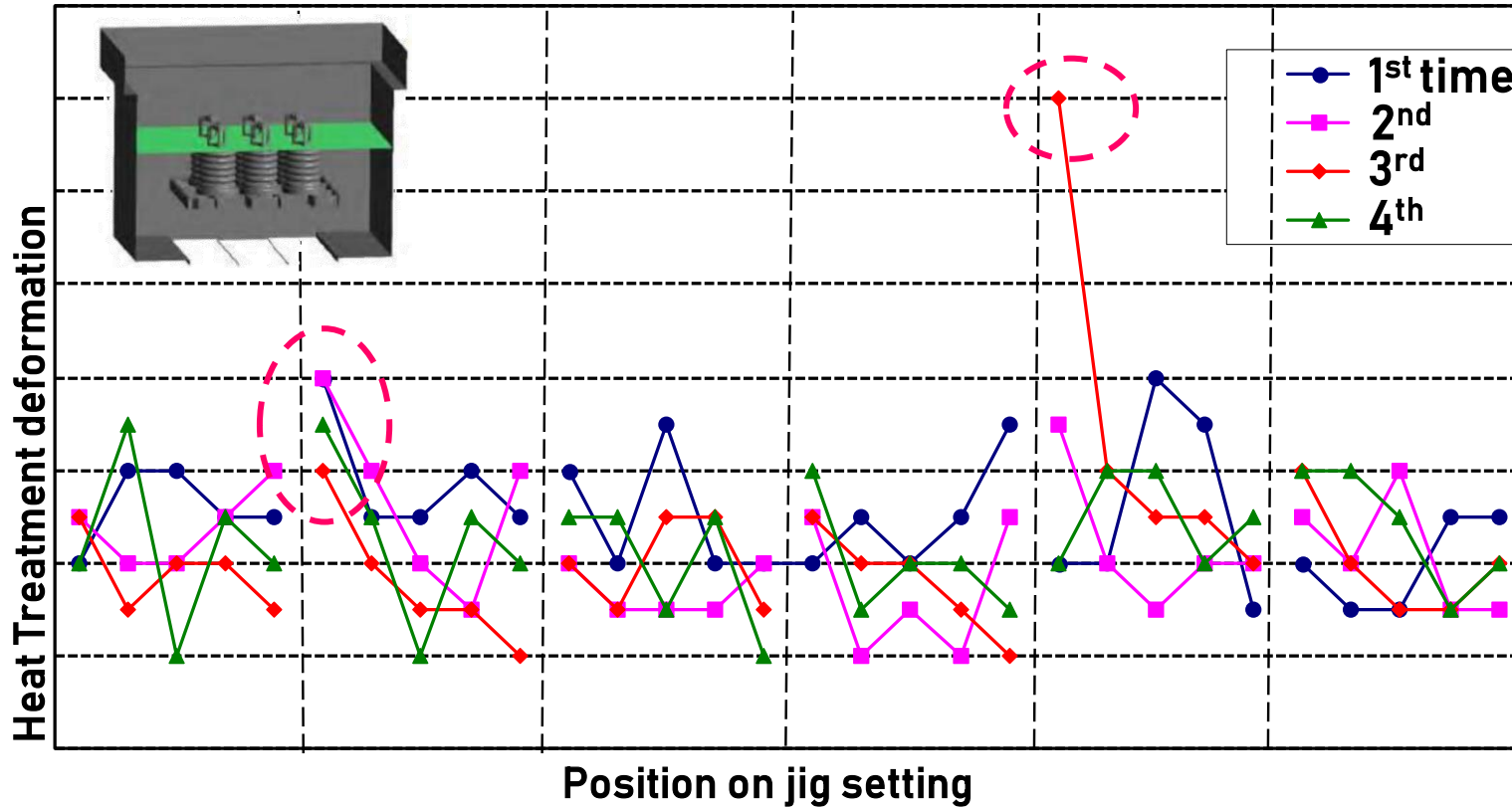
Schematic of Vapor Blanket Collapse

Configuration of Heat Treatment Simulation -Overall Flow-



Uneven cooling conditions on the surface affect heat treatment distortion
Influence of thermal boundary conditions on the results of heat treatment simulation, Tsuyoshi Sugimoto, Dong Ying Ju, Materials Transactions 59(6) 950-956 2018

Variation of Heat Treatment Deformation



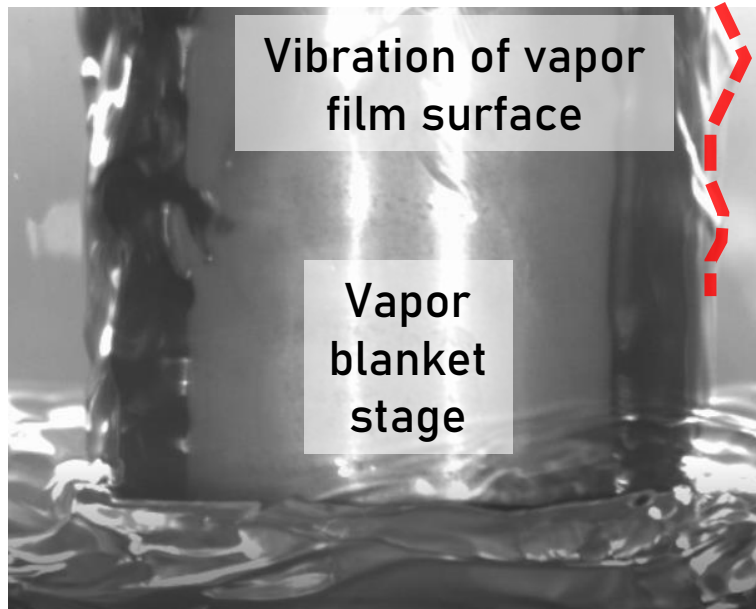
In mass production load setting and iterative quenching, at the specific position

- ✓ The average value of quenching deformation may become large.
- ✓ Suddenly large heat treatment deformation appears due to repeated quenching processing
⇒ "Deformation Variation"

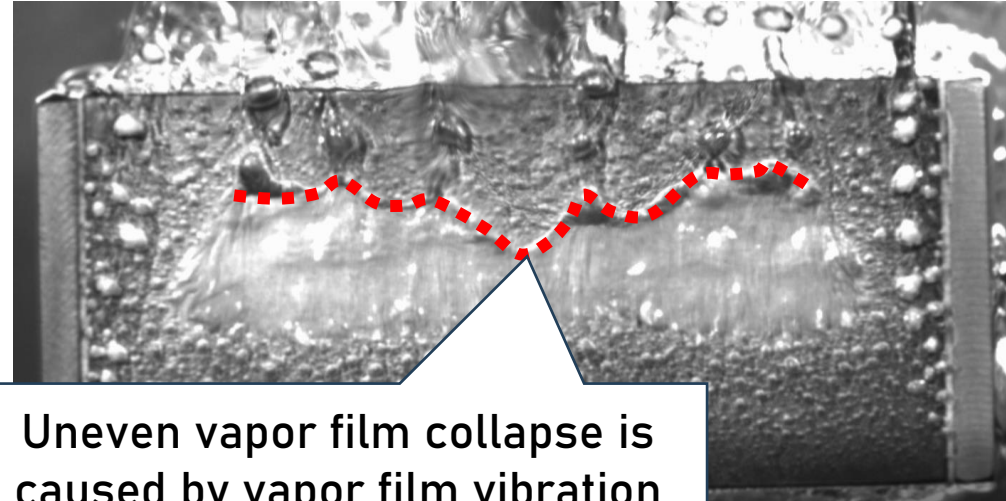


The simulation method is required to reproduce repeated variations in oil quenching process.

Variation of Cooling



Vapor film vibration



Cooling variation

It is seemed that the variation in vapor film collapse is caused by the vibration of the vapor film.



This is one of the reason of variation of quenching deformation variation

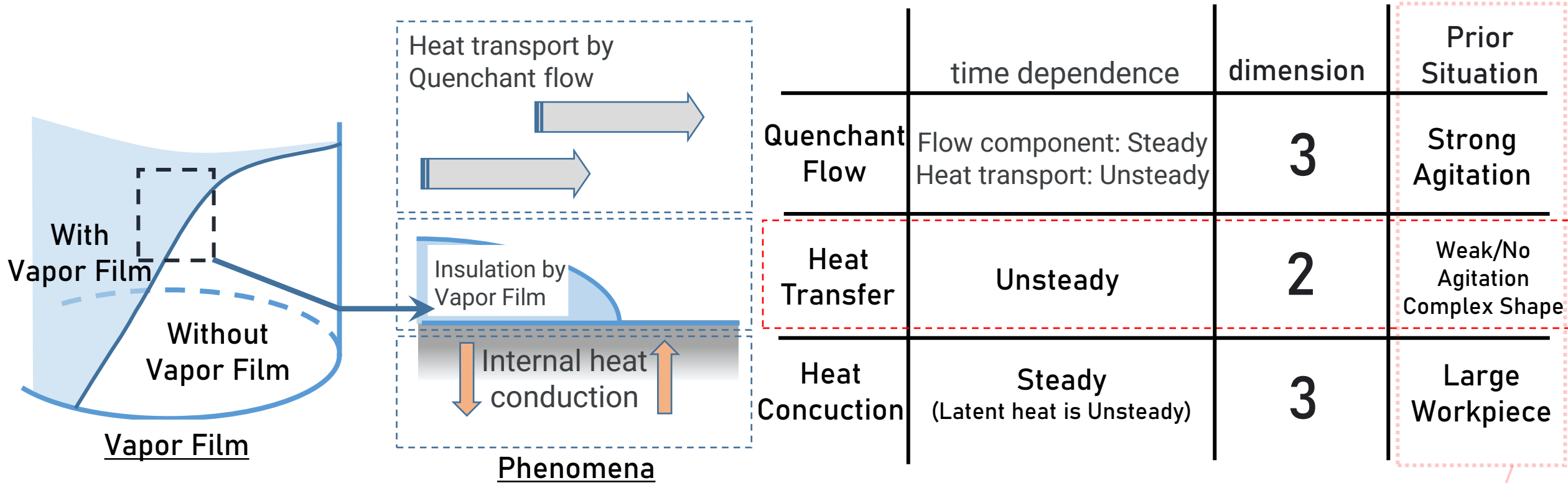
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Low dimension cellular automaton

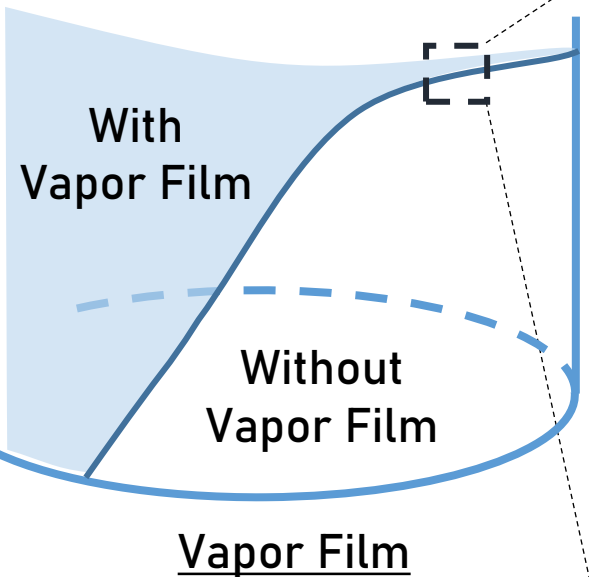


Analysis of thermal insulation by vapor film allows prediction of specific cooling conditions with low load (two-dimensional)

Calculation method between heat treatment simulation and computer fluid dynamics, Tsuyoshi Sugimoto, Kouichi Taniguchi, Shigenori Yamada, Toshiyuki Matsuno, Masaru Sonobe, Dong Ying Ju, Materials Performance and Characterization, 2018, 8(2) 37-49



Basic Equation



Phase transformation model by Wolfram

Perspective Cell

	S_4^t	
S_2^t	S_0^t	S_1^t
	S_3^t	

Phase Change:
Von Neumann Neighborhood
(Weakly affected by surroundings cells)

T_8^t	T_4^t	T_5^t
T_2^t	T_0^t	T_1^t
T_3^t	T_3^t	T_7^t

Temperature:
Moore Neighborhood
(Strongly affected by surroundings cells)

S_i^t : Phase
0: Vapor Blanket Stage t :time
1: Boiling Stage i :Position
2: Convection Stage
 T_i^t : Temperature

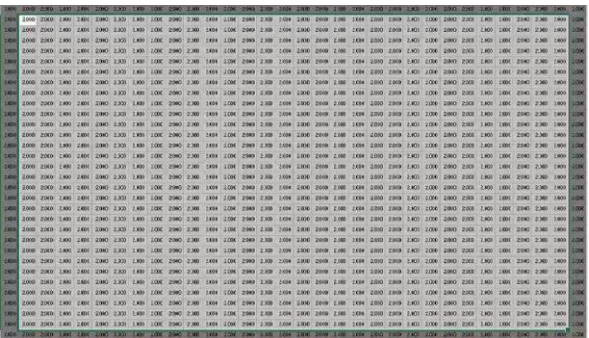
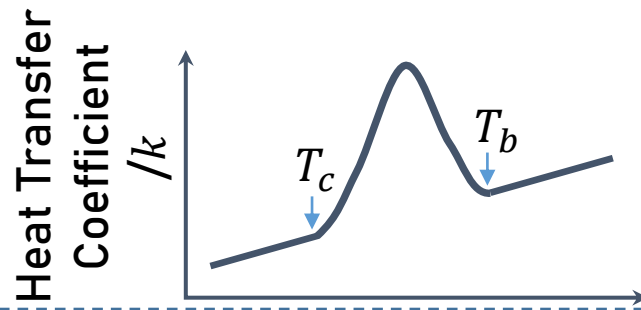
Wolfram, S., *A New Kind of Science*,
Wolfram Store, 2007

Sub-cool boiling model

Phase Change : $S_0^t = 0$ and $\sum_i S_i^t \geq b$ and $T_0^t \leq T_b$ then $S_0^{t+1} = 1$, $T_0^{t+1} = T_0^t - \alpha$..Eq. (1)
 $S_0^t = 1$ and $\sum_i S_i^t \geq c$ and $T_0^t \leq T_c$ then $S_0^{t+1} = 2$, $T_0^{t+1} = T_0^t$.. Eq.(2)

Temperature : $T_0^{t+1} = T_0^t + \left\{ \frac{1}{6} (T_1^t + T_2^t + T_3^t + T_4^t) + \frac{1}{12} (T_5^t + T_6^t + T_7^t + T_8^t) - k \cdot (T_e - T_0^t) \right\}$.. Eq.(3)

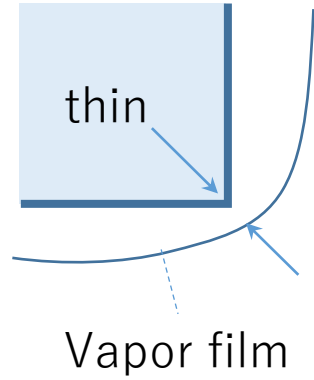
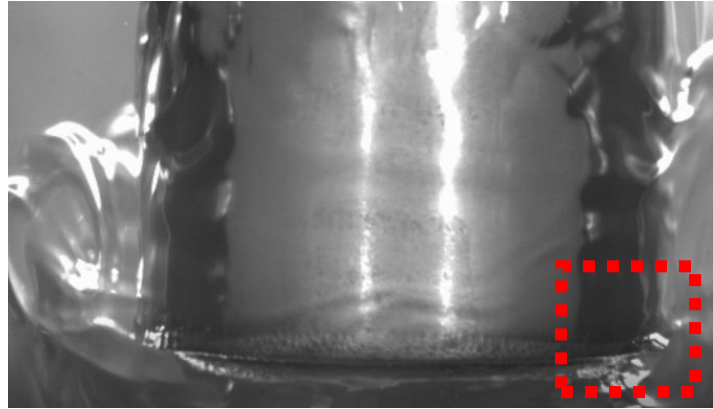
α : Latent Heat
 b, c : Shape Factor
 T_e : Quenchant Temperature
 k : Heat Transfer Coefficient



Vapor film collapse using cellular automaton method

Formulation of Sharp Edge Shape and Tilt of Vapor Film

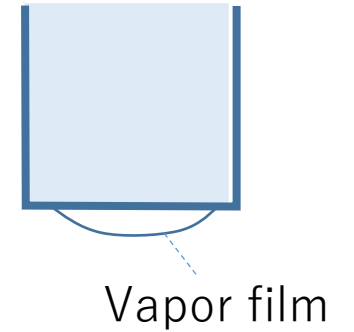
Sharp edge and tilt is reflected as position change of vapor film thickness.



$$b = b_0 + \delta \quad \cdot \cdot \text{Eq. (5)}$$

δ : decrease of vapor film thickness due to sharp edge

Effect of Edge Shape



$$b = b_0 - \varepsilon \cos \theta \quad | \theta \leq 0 \quad \cdot \cdot \text{Eq. (6)}$$

θ : Tilt of surface

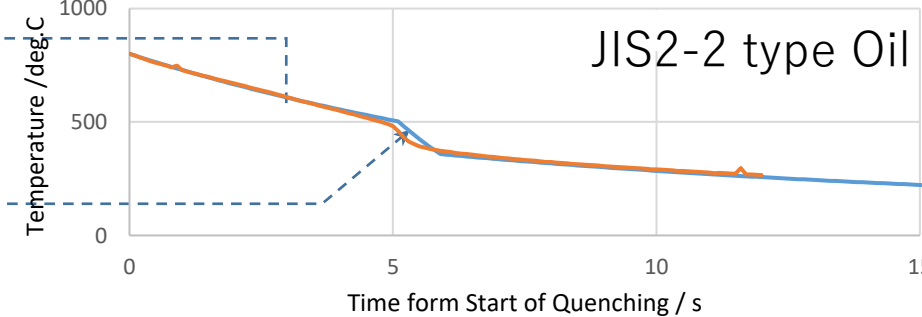
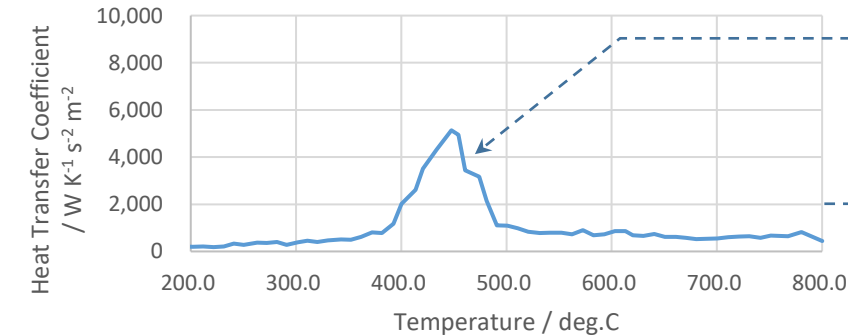
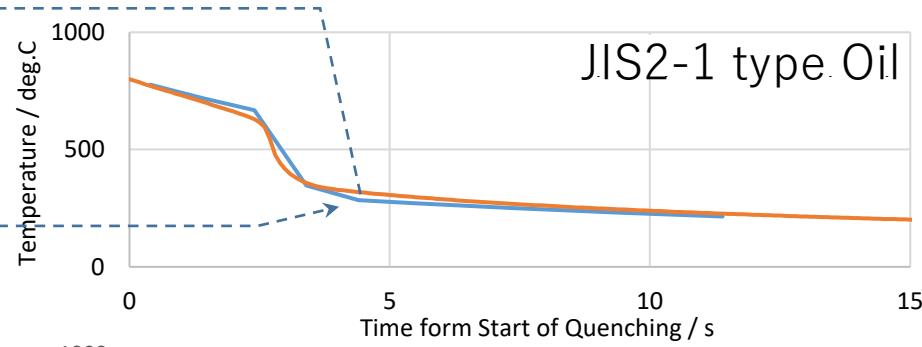
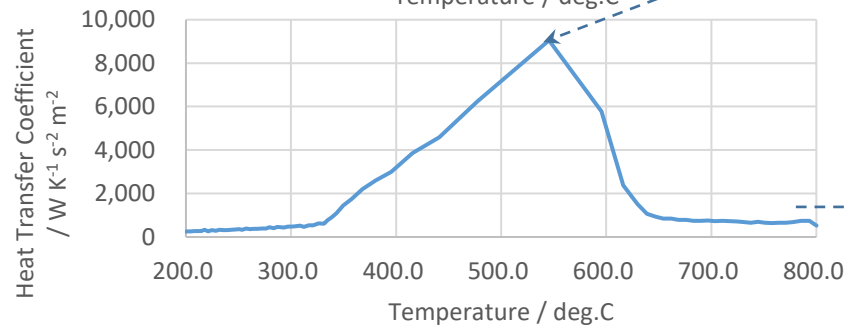
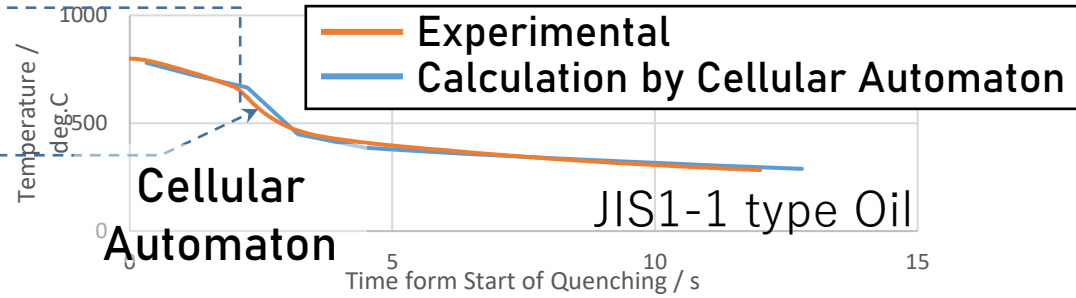
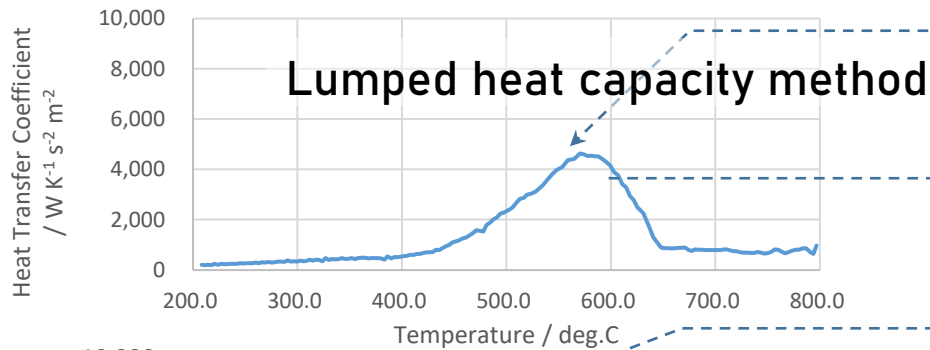
(Horizontal:0, downward is minus)

Effect of Surface Tilt

Verification by JIS silver bar prove



Characteristics of oil and each cooling curve is be able to reproduce



Measured Point

JIS K2422 type A
Silver Probe

Experimental cooling curve and lumped heat capacity calculation:
<https://www.jsht.or.jp/study/>



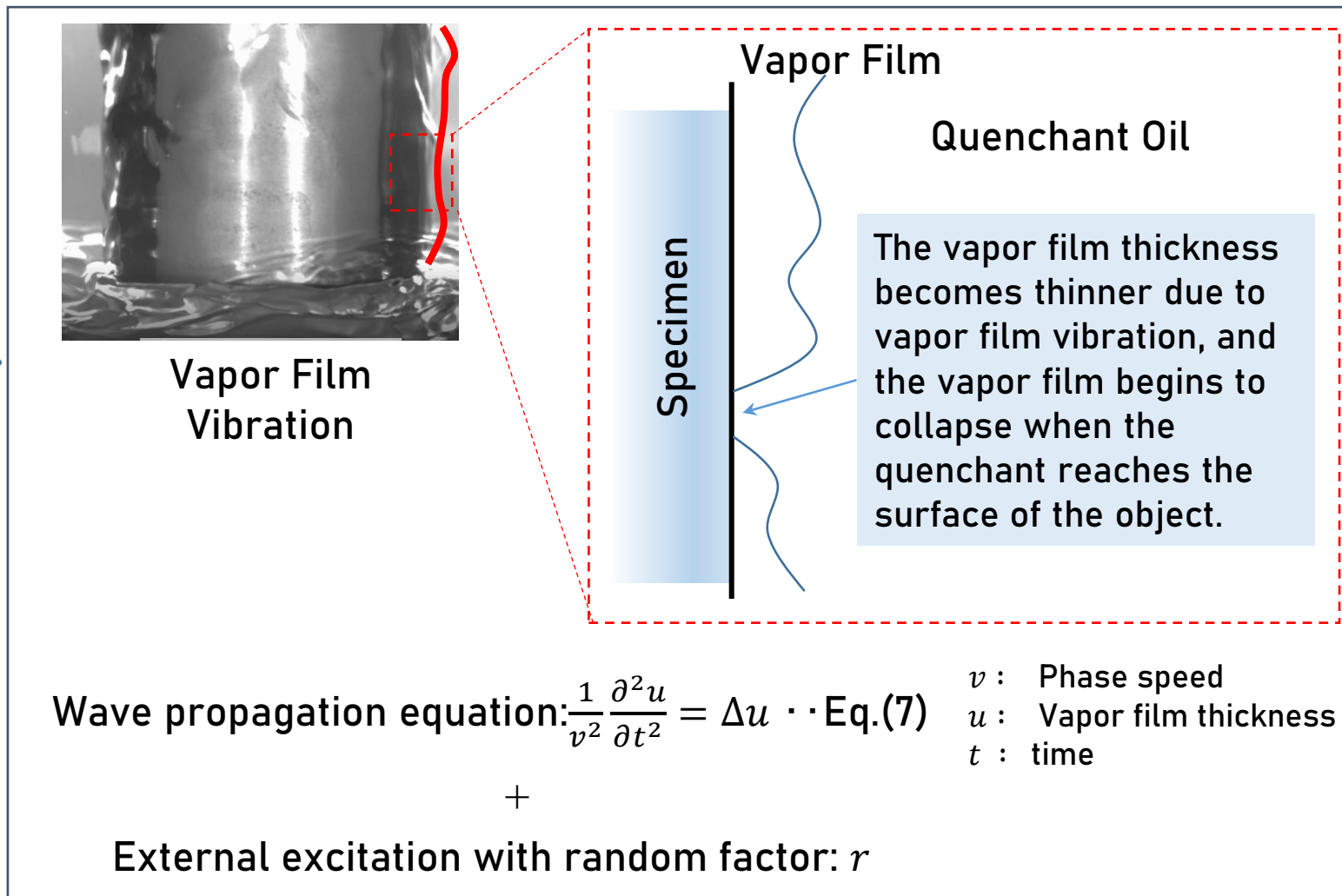
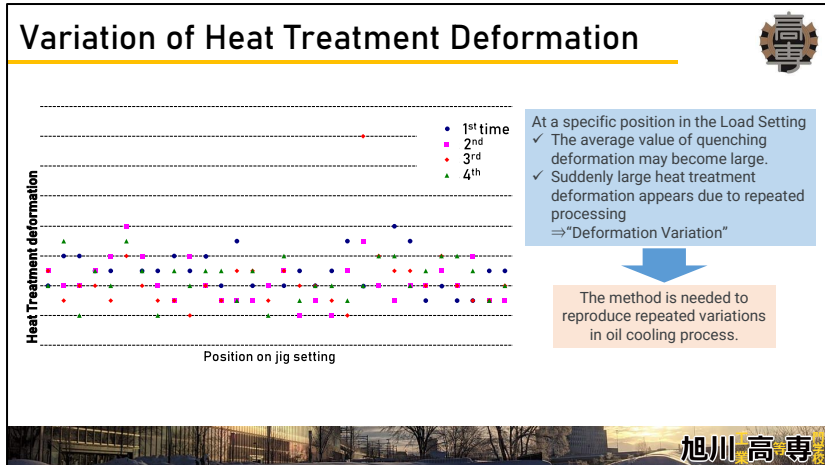


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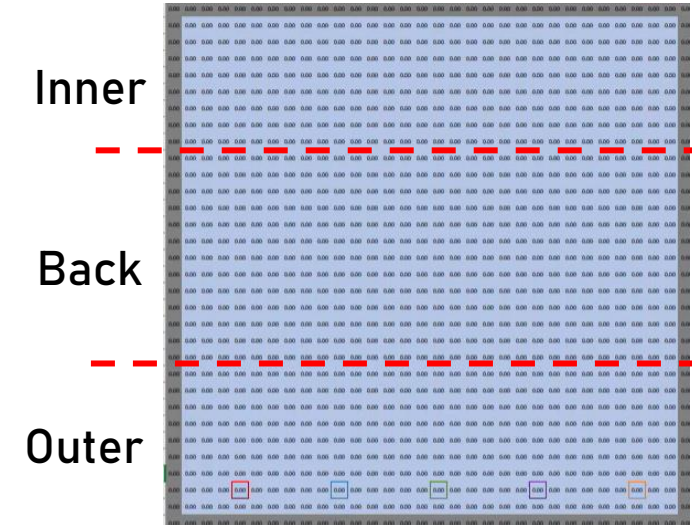
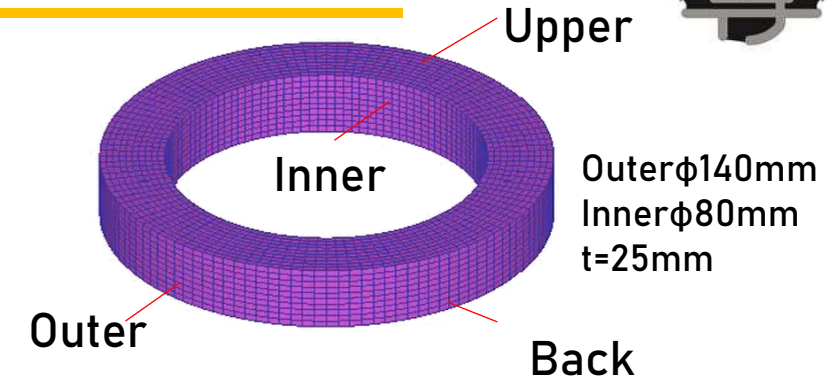
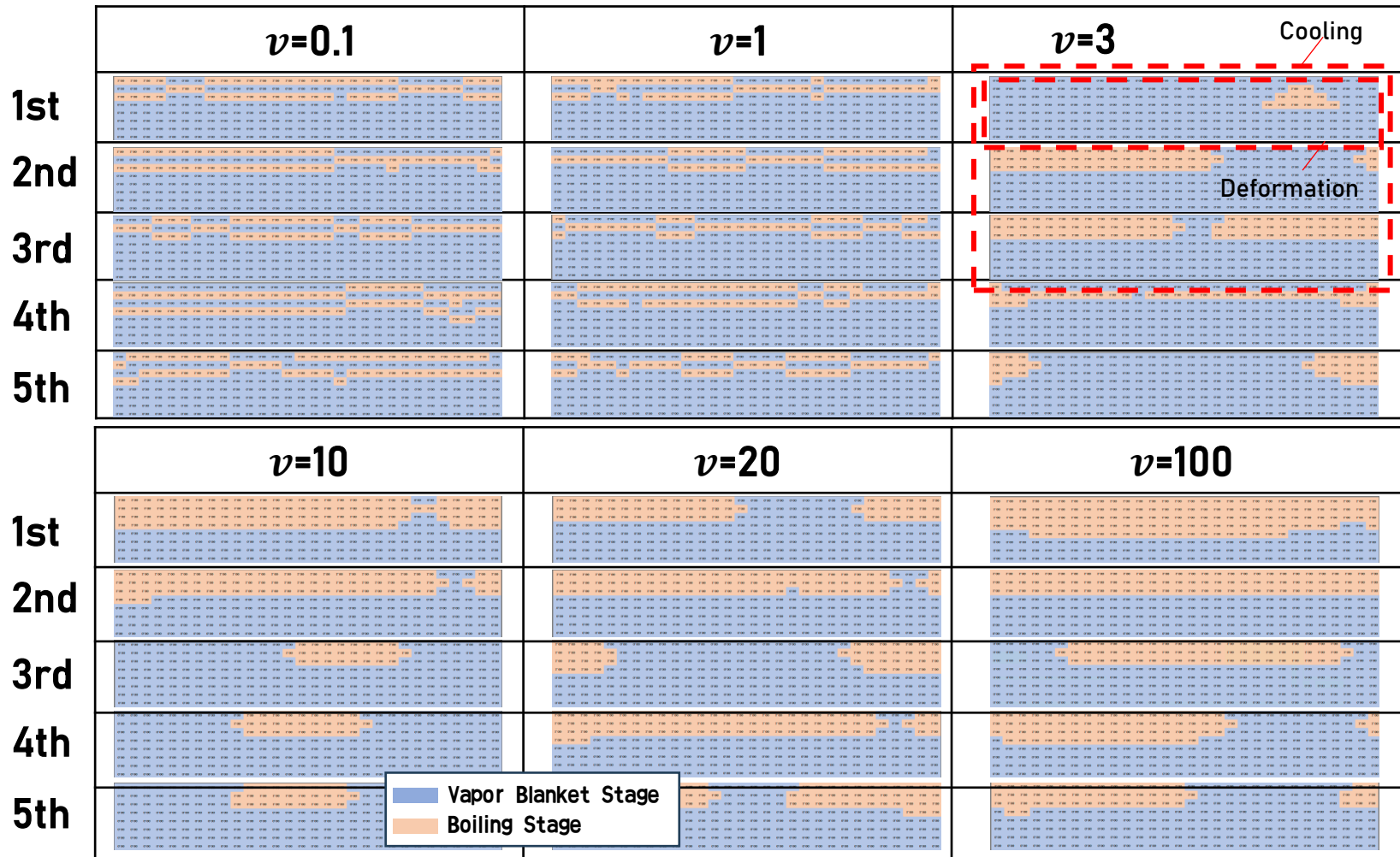
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Cooling variations due to vapor film vibration



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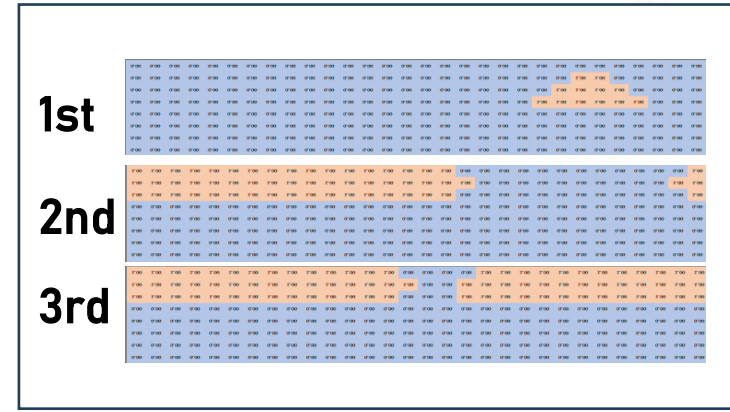
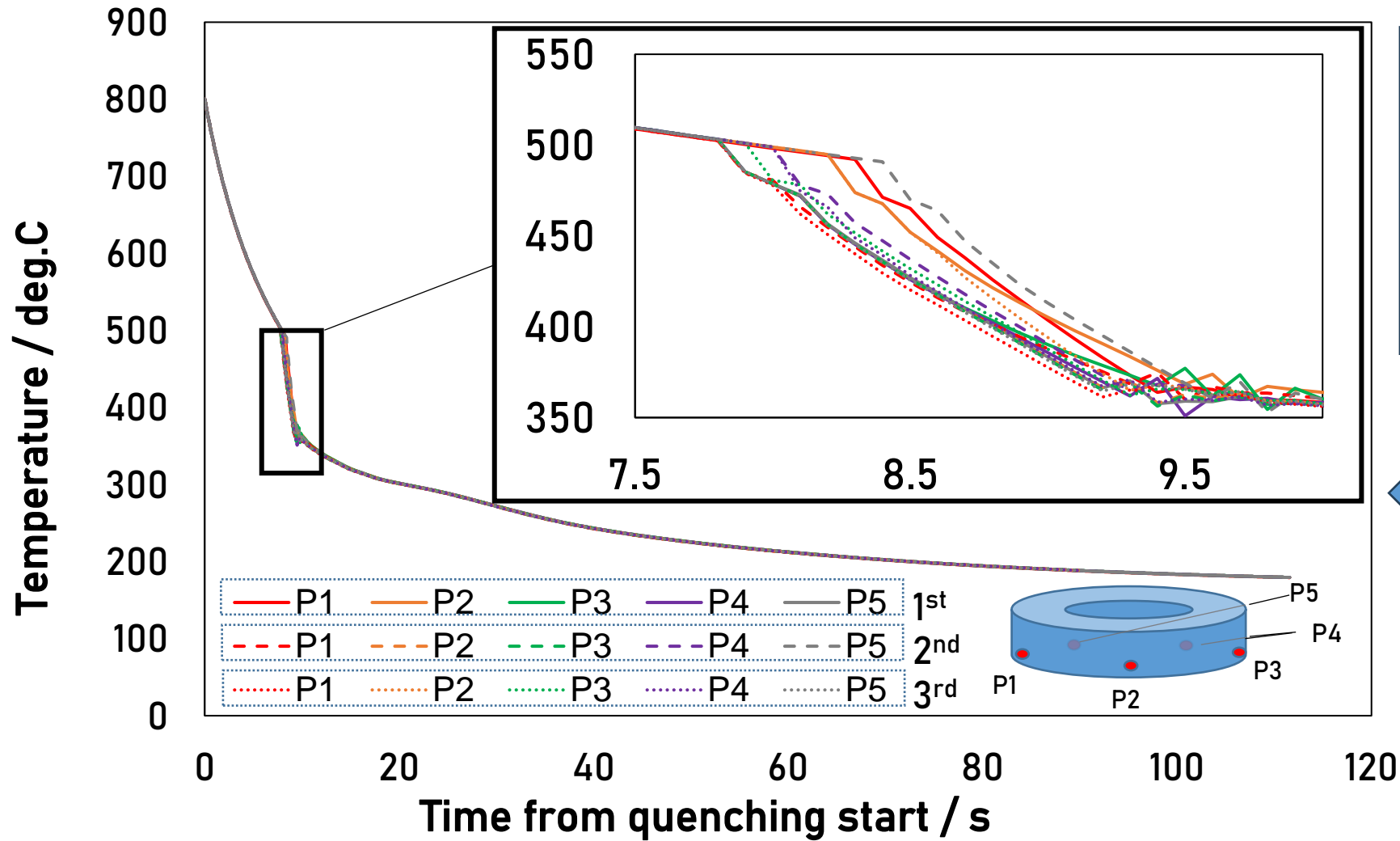
Vapor film collapses at $v=3$

The vapor film collapse form/repetitive change depending on the phase velocity

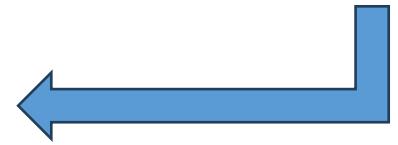
Repeated changes in vapor film collapse at the outer periphery of ring when changing phase speed v



Cooling Curve Variation



Vapor Film Collapse Change



Cooling curve is changed due to Vapor film collapse variation

Cooling Curve Change in Phase Velocity $v=3$



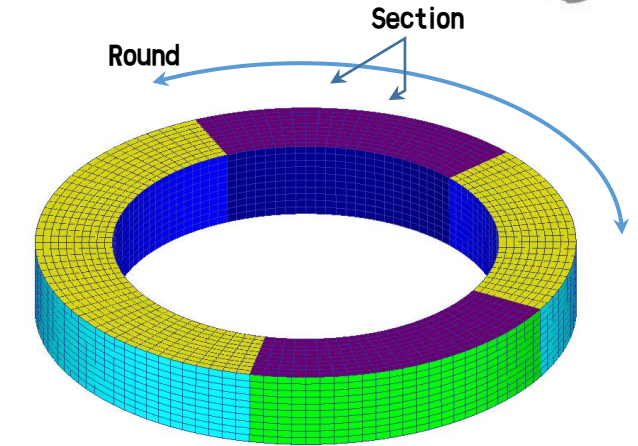
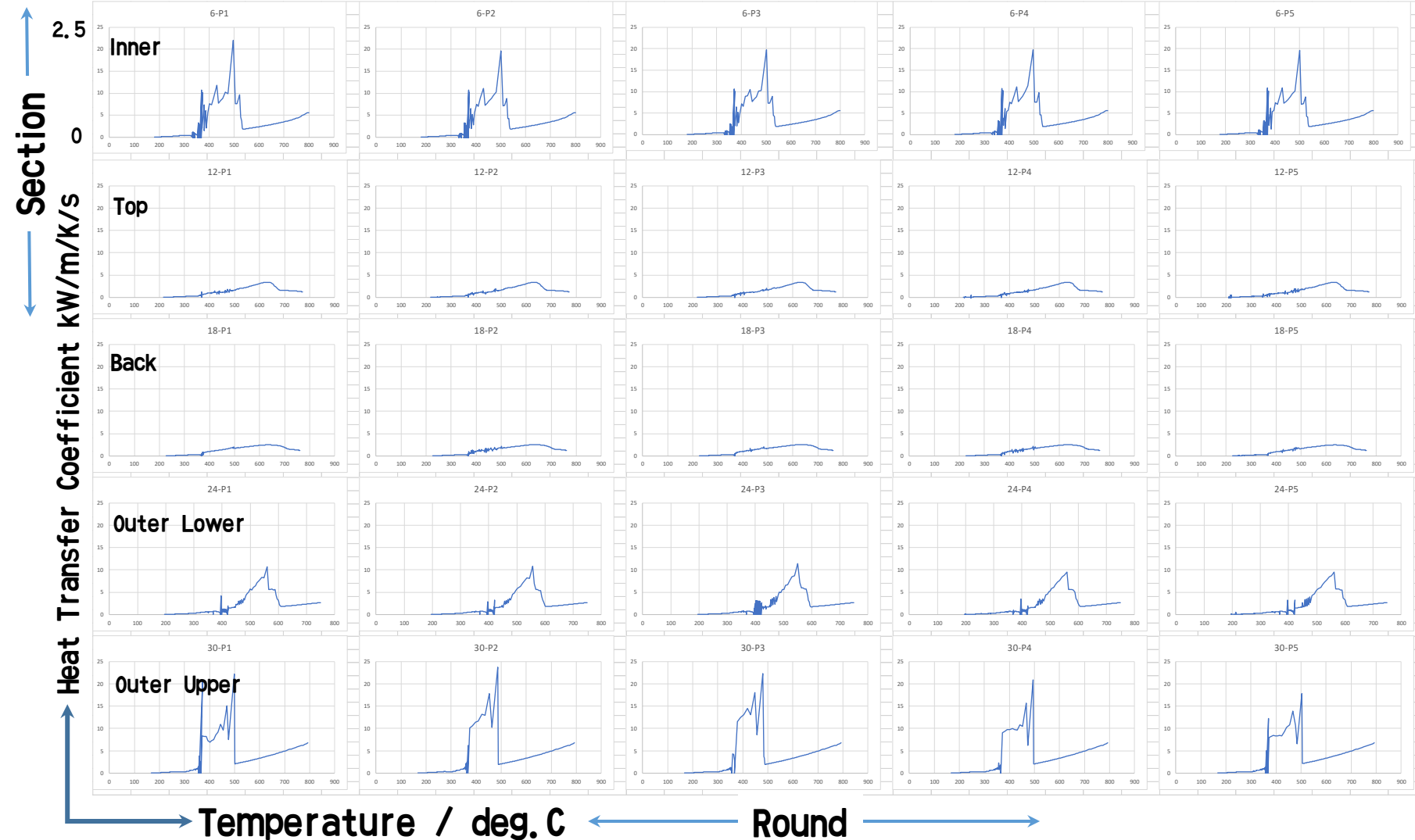


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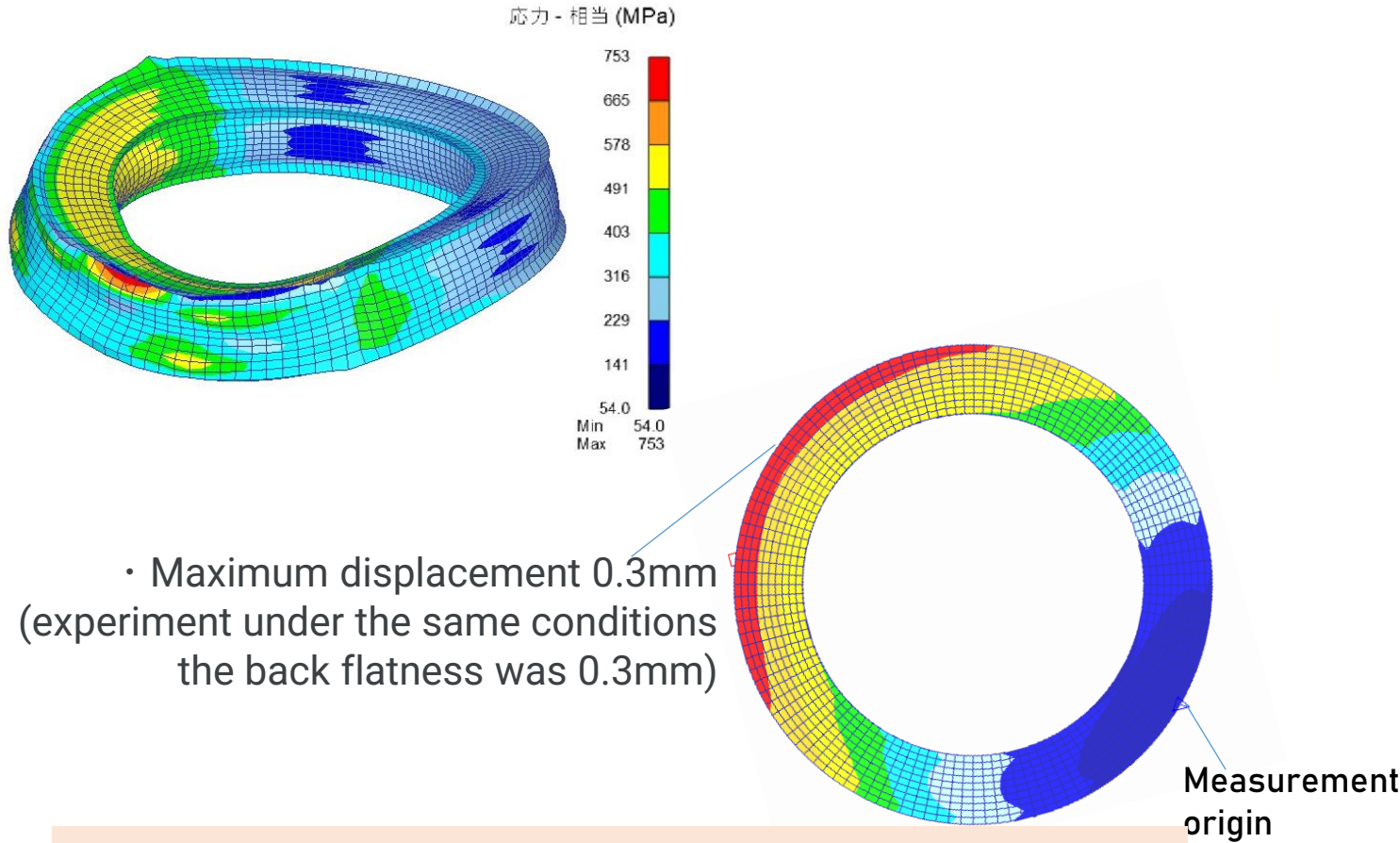
Estimate Heat Transfer Coefficient($\nu=3$)



The surface was divided and a calculated heat transfer coefficient was assigned to each part.

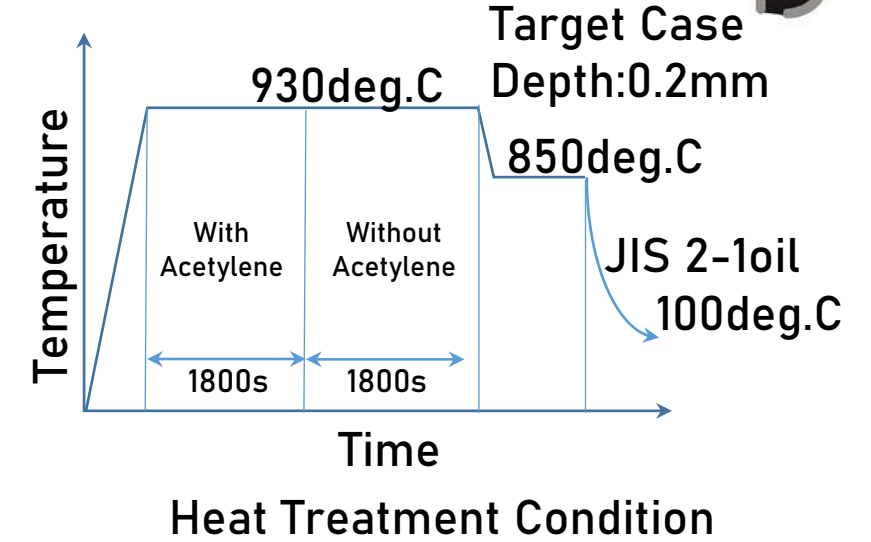
- The heat transfer coefficient on the back side is low
- At the end of boiling stage, Vapor bubbles are generated, which causes the heat transfer coefficient to rise and fall.
- Minimum film boiling temperature is changed

Result of Deformation



- Maximum displacement 0.3mm (experiment under the same conditions the back flatness was 0.3mm)

- Quantitatively, the calculations are mostly accurate.
- Thermal deformation due to initial temperature unevenness affects the final product.



Calculation Condition

Solver	SFTC DEFORM-HT ver13
Nodes	14520
Elements	12000
Coating Mesh Size	0.1mm
Element Shape	Hexagon





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Conclusion

From Results

- ✓ Cooling variations may occur due to vapor film vibration. Heat treatment deformation occurs due to cooling variations.
- ✓ This method is able to analyse repeated variations in heat treatment deformation at low cost.

Next step

We would like to consider mass production load setting. The coupling of the phase velocity v (~oil properties), vapor film excitation force r (~flow velocity, etc.) and steady fluid analysis will realize this one.

Acknowledges

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END

